

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Advanced Subsidiary Level and Advanced Level

CANDIDATE NAME			
CENTRE NUMBER		CANDIDATE NUMBER	
PHYSICS		9702	2/23
Paper 2 AS St	ructured Questions	October/November 2	2012

Candidates answer on the Question Paper.

No Additional Materials are required.

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in. Write in dark blue or black pen. You may use a soft pencil for any diagrams, graphs or rough working.

Do not use staples, paper clips, highlighters, glue or correction fluid. DO **NOT** WRITE IN ANY BARCODES.

Answer all questions.

You may lose marks if you do not show your working or if you do not use appropriate units.

At the end of the examination, fasten all your work securely together.

The number of marks is given in brackets [] at the end of each question or part question.

For Exam	iner's Use
1	
2	
3	
4	
5	
6	
Total	

This document consists of **12** printed pages.



1 hour

Data

speed of light in free space,	$c = 3.00 \times 10^8 \mathrm{ms^{-1}}$
permeability of free space,	$\mu_0 = 4\pi \times 10^{-7} \mathrm{Hm^{-1}}$
permittivity of free space,	$\varepsilon_0 = 8.85 \times 10^{-12} \mathrm{F}\mathrm{m}^{-1}$
	$(\frac{1}{4\pi\varepsilon_0} = 8.99 \times 10^9 \mathrm{mF^{-1}})$
elementary charge,	$e = 1.60 \times 10^{-19} \mathrm{C}$
the Planck constant,	$h = 6.63 \times 10^{-34} \mathrm{Js}$
unified atomic mass constant,	$u = 1.66 \times 10^{-27} \text{ kg}$
rest mass of electron,	$m_{\rm e} = 9.11 \times 10^{-31} \rm kg$
rest mass of proton,	$m_{\rm p} = 1.67 \times 10^{-27} \rm kg$
molar gas constant,	$R = 8.31 \mathrm{JK^{-1}mol^{-1}}$
the Avogadro constant,	$N_{\rm A} = 6.02 \times 10^{23} {\rm mol}^{-1}$
the Boltzmann constant,	$k = 1.38 \times 10^{-23} \mathrm{J K^{-1}}$
gravitational constant,	$G = 6.67 \times 10^{-11} \mathrm{N}\mathrm{m}^2\mathrm{kg}^{-2}$
acceleration of free fall,	$g = 9.81 \text{ ms}^{-2}$

2

Formulae

uniformly accelerated motion,	$s = ut + \frac{1}{2}at^2$ $v^2 = u^2 + 2as$
work done on/by a gas,	$W = \rho \Delta V$
gravitational potential,	$\phi = -\frac{Gm}{r}$
hydrostatic pressure,	$p = \rho g h$
pressure of an ideal gas,	$p = \frac{1}{3} \frac{Nm}{V} < c^2 >$
simple harmonic motion,	$a = -\omega^2 x$
velocity of particle in s.h.m.,	$v = v_0 \cos \omega t$ $v = \pm \omega \sqrt{(x_0^2 - x^2)}$
electric potential,	$V = \frac{Q}{4\pi\varepsilon_0 r}$
capacitors in series,	$1/C = 1/C_1 + 1/C_2 + \dots$
capacitors in parallel,	$C = C_1 + C_2 + \dots$
energy of charged capacitor,	$W = \frac{1}{2}QV$
resistors in series,	$R = R_1 + R_2 + \dots$
resistors in parallel,	$1/R = 1/R_1 + 1/R_2 + \dots$
alternating current/voltage,	$x = x_0 \sin \omega t$
radioactive decay,	$x = x_0 \exp(-\lambda t)$
decay constant,	$\lambda = \frac{0.693}{t_{\frac{1}{2}}}$

4

			Answer all t	he questions in the	spaces provided	1.	1	For Examiner's Use
1	(a)	The spacing b	etween two ato	ms in a crystal is 3	.8 × 10 ^{−10} m. Stat	te this distance	in pm.	
				spacir	ng =		pm [1]	
	(b)	Calculate the	time of one day	in Ms.				
				tin	1e =		Ms [1]	
	(c)	The distance f to travel from t	rom the Earth to the Sun to the E	o the Sun is 0.15 Ti Earth.	n. Calculate the t	ime in minutes	for light	
				tin	1e =		min [2]	
	(d)	Underline all t	he vector quant	ities in the list belo	W.			
		distance	energy	momentum	weight	work	[1]	

(e) The velocity vector diagram for an aircraft heading due north is shown to scale in Fig. 1.1. There is a wind blowing from the north-west. Examiner's





The speed of the wind is 36 m s^{-1} and the speed of the aircraft is 250 m s^{-1} .

- Draw an arrow on Fig. 1.1 to show the direction of the resultant velocity of the (i) aircraft. [1]
- Determine the magnitude of the resultant velocity of the aircraft. (ii)

resultant velocity = ms^{-1} [2]

For

Use

2 Two planks of wood AB and BC are inclined at an angle of 15° to the horizontal. The two wooden planks are joined at point B, as shown in Fig. 2.1.

0.26 m 15° B 15° 0.26 m 0.26 m



A small block of metal M is released from rest at point A. It slides down the slope to B and up the opposite side to C. Points A and C are 0.26m above B. Assume frictional forces are negligible.

(a) (i) Describe and explain the acceleration of M as it travels from A to B and from B to C.

[3]

(ii) Calculate the time taken for M to travel from A to B.

time = s [3]

(iii) Calculate the speed of M at B.

speed = ms^{-1} [2]

(b) The plank BC is adjusted so that the angle it makes with the horizontal is 30°. M is released from rest at point A and slides down the slope to B. It then slides a distance along the plank from B towards C.

Use the law of conservation of energy to calculate this distance. Explain your working.

For Examiner's Use



(ii) When the cyclist is moving at a constant speed of 12 m s^{-1} the resistive force is 48 N. Show that the power of the cyclist is about 600 W. Explain your working.

For Examiner's Use

[2]

(iii) Use Fig. 3.1 to show that the acceleration of the cyclist when his speed is 8.0 m s^{-1} is about 0.5 m s^{-2} .

[2]

(iv) The total mass of the cyclist and bicycle is 80 kg. Calculate the resistive force *R* acting on the cyclist when his speed is 8.0 m s^{-1} . Use the value for the acceleration given in (iii).

R = N [3]

(v) Use the information given in (ii) and your answer to (iv) to show that, in this situation, the resistive force *R* is proportional to the speed *v* of the cyclist.

[1]

4 A circuit used to measure the power transfer from a battery is shown in Fig. 4.1. The power is transferred to a variable resistor of resistance *R*.



Fig. 4.1

The battery has an electromotive force (e.m.f.) E and an internal resistance r. There is a potential difference (p.d.) V across R. The current in the circuit is I.

(a) By reference to the circuit shown in Fig. 4.1, distinguish between the definitions of e.m.f. and p.d.



(b) Using Kirchhoff's second law, determine an expression for the current *I* in the circuit.

[1]

For

Examiner's Use

For

(c) The variation with current I of the p.d. V across R is shown in Fig. 4.2.



efficiency = % [2]

romagnetic spectrum are represented by blocks	n regions of the electromagne Fig. 5.1.	The seven r A to G in Fig				
visible region	visib					
↓						
C D E F G	в С	А				
	th decreasing —	wavelength				
Fig. 5.1	Fi					
ible region D is 500 nm.	wavelength for the visible region	A typical wa				
ons and give a typical wavelength for each of the	e the principal radiations and and F.	(i) Name t B, E an				
B: name: m						
E: name: m						
wavelength:						
	me:	F: name				
brresponding to a wavelength of 500 nm. frequency =shown in Fig. 5.1 can be polarised. Explain the	me:	F: name (ii) Calcula All the wave of the term <i>j</i>				

(a)	β-ra (i)	diation is emitted during the spontaneous radioactive decay of an unstable nucleus. State the nature of a β -particle.
	(-)	[1]
	(ii)	State two properties of β -radiation.
		1
		2[2]
	(iii)	Explain the meaning of spontaneous radioactive decay.
		[1]
(b)	The of a	following equation represents the decay of a nucleus of hydrogen-3 by the emission $\beta\mbox{-}particle.$
	Con	nplete the equation.
		$^{3}_{1}H \rightarrow \dots He + \dots \beta$ [2]
(c)	The	β -particle is emitted with an energy of 5.7 × 10 ³ eV.
	Calo	culate the speed of the β -particle.
		speed = $m s^{-1}$ [3]
(d)	A di diffe	fferent isotope of hydrogen is hydrogen-2 (deuterium). Describe the similarities and rences between the atoms of hydrogen-2 and hydrogen-3.

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (UCLES) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

University of Cambridge International Examinations is part of the Cambridge Assessment Group. Cambridge Assessment is the brand name of University of Cambridge Local Examinations Syndicate (UCLES), which is itself a department of the University of Cambridge.