

Cambridge International Examinations

Cambridge International Advanced Subsidiary and Advanced Level

CHEMISTRY 9701/31

Paper 3 Advanced Practical Skills 1

May/June 2018

CONFIDENTIAL INSTRUCTIONS



Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.

The Supervisor's attention is drawn to the Supervisor's Report on page 7 which must be completed and returned with the scripts.

If you have any queries regarding these Confidential Instructions, please contact Cambridge stating the Centre number, the nature of the query and the syllabus number quoted above.

email info@cie.org.uk phone +44 1223 553554 fax +44 1223 553558

This document consists of 8 printed pages.



Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution.

Only those tests described in the Question Paper should be attempted.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out. The following codes are used where relevant.

C corrosive
HH health hazard
F flammable
MH moderate hazard
T acutely toxic
O oxidising

N hazardous to the aquatic environment

The attention of Supervisors is drawn to any local regulations relating to safety and first aid.

Hazard Data Sheets should be available from your chemical suppliers.

Before the examination

1 Access to the Question Paper is NOT permitted in advance of the examination.

2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the Question Paper to be completed.

In preparing materials, the bulk quantity for each substance should be increased by 25% as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate.

If the concentrations differ slightly from those specified, the Examiners will make the necessary allowance. They should be informed of the exact concentrations.

3 Labelling of materials

Materials must be labelled as specified in these Confidential Instructions. Materials with an **FA** code number should be so labelled **without** the identities being included on the label. Where appropriate the identity of an **FA** coded chemical is given in the Question Paper itself.

4 Identity of materials

It should be noted that descriptions of materials given in the Question Paper may not correspond with the specifications in these Confidential Instructions. **The candidates must assume the descriptions given in the Question Paper.**

5 Size of group

In view of the difficulty of the preparation of large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

Apparatus

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- 2 Pipette fillers (or equivalent safety devices), suitable eye protection and disposable gloves should be used where necessary.
- 3 For each candidate
 - $2 \times 50 \, \text{cm}^3$ burette
 - $1 \times 25 \, \text{cm}^3$ pipette
 - $1 \times \text{thermometer} (-10 ^{\circ}\text{C} \text{ to } +110 ^{\circ}\text{C at } 1 ^{\circ}\text{C})$
 - 2 × burette stand and clamp
 - 1 × funnel (for filling burette)
 - 1 × white tile
 - 2 × 150 cm³ or 250 cm³ conical flask
 - 1 × 250 cm³ beaker
 - 1 × Bunsen burner
 - 1 × heatproof mat
 - $1 \times tripod$
 - 1 × gauze
 - 1 × test-tube holder
 - 7 × test-tube*
 - 3 × boiling tube
 - 1 × spatula
 - 1 × test-tube rack
 - 2 × teat/dropping pipette
 - 1 × wash bottle containing distilled water
 - 1 × pen for labelling glassware

paper towels

^{*}Candidates are expected to rinse and reuse test-tubes and boiling tubes where possible. Additional tubes should be available.

Chemicals required

- It is especially important that great care is taken that the confidential information given below does not reach the candidates either directly or indirectly.
- It should be noted that descriptions of substances given in the Question Paper may not correspond with the specifications in these Confidential Instructions.

3 Particular requirements

FA 1 150 cm³ sodium carbonate sodium carbonate Dissolve 5.20 g anhydrous Na ₂ CO ₂ [MH] in each dm³ of solution. FA 2 150 cm³ acid 0.100 moldm³ hydrochloric [MH] Prepare 2.0 moldm³ of solution. This is the sacid master and master	hazard	label	per candidate	identity	notes (hazards given in this column are for the raw materials)
FA 2 150 cm³ 0.100 mol dm⁻³ hydrochloric acid methyl orange 5 cm³ methyl orange indicator FA 3 120 cm³ 0.100 mol dm⁻³ sodium FA 4 150 cm³ 0.100 mol dm⁻³ hydrochloric acid acid bromophenol blue 5 cm³ bromophenol blue indicator FA 5 10 cm³ 2.0 mol dm⁻³ methanoic acid FA 7 1g zinc carbonate FA 8 10 cm³ 0.2 mol dm⁻³ copper(II) nitrate sodium carbonate 1g anhydrous sodium carbonate distilled water 300 cm³ anhydrous sodium carbonate		FA 1	150 cm ³	0.049 mol dm ⁻³ sodium carbonate	Dissolve 5.20g anhydrous Na ₂ CO ₃ [MH] or 14.0g Na ₂ CO ₃ .10H ₂ O [MH] in each dm³ of solution.
methyl orange 5cm³ methyl orange indicator FA3 120cm³ 0.100 mol dm⁻³ sodium FA4 150cm³ 0.100 mol dm⁻³ hydrochloric bromophenol blue 5cm³ bromophenol blue indicator FA5 10cm³ 2.0 mol dm⁻³ methanoic acid FA7 1g zinc carbonate FA8 10cm³ 0.2 mol dm⁻³ copper(II) nitrate sodium carbonate 1g anhydrous sodium carbonate distilled water 300 cm³		FA 2	150 cm ³	0.100 mol dm ⁻³ hydrochloric acid	Prepare 2.0 mol dm ⁻³ HC <i>l</i> using the instructions in the current syllabus. Dilute 50 cm ³ in each dm ³ of solution. This is the same as FA 4 .
FA 3120 cm³ hydroxide0.100 mol dm⁻³ sodium hydroxideFA 4150 cm³0.100 mol dm⁻³ hydrochloric acidbromophenol blue5 cm³bromophenol blue indicatorFA 510 cm³2.0 mol dm⁻³ methanoic acidFA 71 gzinc carbonateFA 810 cm³0.2 mol dm⁻³ copper(II) nitratesodium carbonate1 ganhydrous sodium carbonatedistilled water300 cm³anhydrous sodium carbonate	[N][F] [MH] [HH]	methyl orange	5 cm³	methyl orange indicator	See preparation instructions in the current syllabus.
FA 4150 cm³ acid0.100 mol dm⁻³ hydrochloric acidbromophenol blue5 cm³bromophenol blue indicatorFA 510 cm³2.0 mol dm⁻³ methanoic acidFA 71 gzinc carbonateFA 810 cm³0.2 mol dm⁻³ copper(II) nitratesodium carbonate1 ganhydrous sodium carbonatedistilled water300 cm³anhydrous sodium carbonate		FA 3	120 cm ³	0.100 mol dm ⁻³ sodium hydroxide	Dissolve $4.00\mathrm{g}$ NaOH [C] in each dm 3 of solution.
bromophenol blue5 cm³bromophenol blue indicatorFA 510 cm³2.0 moldm⁻³ methanoic acidFA 71gzinc carbonateFA 810 cm³0.2 moldm⁻³ copper(II) nitratesodium carbonate1ganhydrous sodium carbonatedistilled water300 cm³anhydrous sodium carbonate		FA 4	150 cm ³	0.100 mol dm ⁻³ hydrochloric acid	Prepare 2.0 mol dm ⁻³ HC <i>l</i> using the instructions in the current syllabus. Dilute 50 cm ³ in each dm ³ of solution. This is the same as FA 2 .
FA 510 cm³2.0 moldm⁻³ methanoic acidFA 71gzinc carbonateFA 810 cm³0.2 moldm⁻³ copper(II) nitratesodium carbonate1ganhydrous sodium carbonatedistilled water300 cm³	[F][MH] [HH]	bromophenol blue	5 cm³		See preparation instructions in the current syllabus.
FA 71gzinc carbonateFA 810 cm³0.2 moldm-³ copper(II) nitratesodium carbonate1ganhydrous sodium carbonatedistilled water300 cm³	[MH]	FA 5	10 cm³		Dilute 100 g concentrated HCOOH [C] (85-90%) in each dm 3 of solution.
FA 810 cm³0.2 moldm³ copper(II) nitratesodium carbonate1ganhydrous sodium carbonatedistilled water300 cm³		FA 7	19	zinc carbonate	1 g ± 0.1 g ZnCO ₃ in a stoppered container. Basic zinc carbonate is suitable.
sodium carbonate1ganhydrous sodium carbonatedistilled water300 cm³	[]	FA 8	10 cm ³		Dissolve $48.3\mathrm{g}~\mathrm{Cu(NO_3)_2.3H_2O}$ [C][MH][N] in each dm³ of solution.
	[MH]	sodium carbonate	1g	anhydrous sodium carbonate	1g \pm 0.1g anhydrous Na ₂ CO ₃ [MH] in a stoppered container.
		distilled water	300 cm ³		

reagents. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the The reagents below should also be provided. Unless otherwise stated, each candidate should require no more than 10 cm³ of any of these fact that such an arrangement may lead to contamination of reagents and enhance the opportunity for malpractice between candidates.

hazard		<u>[</u>	[MH]	[C][MH]	<u>ට</u>				[MH]	[MH]
label	dilute hydrochloric acid	dilute nitric acid	dilute sulfuric acid] aqueous ammonia	aqueous sodium hydroxide	0.1 moldm ⁻³ barium chloride or	0.1 mol dm⁻³ barium nitrate	0.05 moldm ⁻³ silver nitrate	limewater	acidified aqueous potassium manganate(VII)
notes					See identity details and preparation instructions in the current syllabus.	•				

5 The following materials and apparatus should be available.

red and blue litmus papers, aluminium foil for testing nitrate/nitrite, wooden splints and the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide

Responsibilities of the Supervisor during the examination

1 The Supervisor, or other competent chemist, must, out of sight of the candidates, carry out the experiments in Question 1 and Question 2 and complete tables of readings on a spare copy of the Question Paper. This should be labelled 'Supervisor's Results' and show the Centre number and appropriate session/laboratory number.

This should be done for **each session** held and **each laboratory** used in that session, and **each batch** of solutions supplied.

N.B. The Question Paper cover requests the candidate to fill in details of the examination session and the laboratory used for the examination.

It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.

2 The Supervisor must complete the Supervisor's Report on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Supervisor's Report. A copy of the Supervisor's Report must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

After the examination

Each envelope returned to Cambridge must contain the following items.

- 1 The scripts of those candidates specified on the barcode label provided.
- 2 A copy of the Supervisor's Results relevant to the candidates in 1.
- A copy of the Supervisor's Report, including details of any difficulties experienced by candidates (see pages 7 and 8).
- **4** The Attendance Register.
- **5** A Seating Plan for each session/laboratory.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

Colour blindness

With regard to colour blindness it is permissible to advise candidates who request assistance on colours of, for example, precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a Special Consideration Form.

SUPERVISOR'S REPORT

This form must be completed and returned in t Results, the Attendance Register and the seati	
Centre number	Name of Centre
The candidate numbers of candidates attending ea	ach session were:
first session	second session

The Supervisor is required to give details overleaf of any difficulties experienced by particular candidates, giving names and candidate numbers. These should include reference to:

- any general difficulties encountered in preparation of materials;
- difficulties due to faulty apparatus or materials;
- accidents to apparatus or materials;
- assistance with respect to colour blindness.

Other cases of hardship, e.g. illness, temporary disability, should be reported directly to Cambridge on the Special Consideration Form.

Report on any difficulties experienced by candidates.
Declaration (to be signed by the Supervisor)
The preparation of this practical examination has been carried out so as to maintain fully the security of the examination.
Signed
Name (in block capitals) (Supervisor)

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